LABORATORY EXERCISES IN PHYSICAL CHEMISTRY

Published @ 2017 Trieste Publishing Pty Ltd

ISBN 9780649116324

Laboratory exercises in physical chemistry by Frederick H. Getman

Except for use in any review, the reproduction or utilisation of this work in whole or in part in any form by any electronic, mechanical or other means, now known or hereafter invented, including xerography, photocopying and recording, or in any information storage or retrieval system, is forbidden without the permission of the publisher, Trieste Publishing Pty Ltd, PO Box 1576 Collingwood, Victoria 3066 Australia.

All rights reserved.

Edited by Trieste Publishing Pty Ltd. Cover @ 2017

This book is sold subject to the condition that it shall not, by way of trade or otherwise, be lent, re-sold, hired out, or otherwise circulated without the publisher's prior consent in any form or binding or cover other than that in which it is published and without a similar condition including this condition being imposed on the subsequent purchaser.

www.triestepublishing.com

FREDERICK H. GETMAN

LABORATORY EXERCISES IN PHYSICAL CHEMISTRY



E3945kx LABORATORY EXERCISES

IN

PHYSICAL CHEMISTRY.

BY

FREDERICK H. GETMAN, PH.D.,

Fellow by Courtesy of The Johns Hopkins University, Carnegie Research Assistant,

FIRST EDITION.

FIRST THOUSAND.

NEW YORK:

JOHN WILEY & SONS.

LONDON: CHAPMAN & HALL, LIMITED

1904.

PREFACE.

With the growth of a new science between physics and chemistry there has arisen need for a new type of laboratory manual. This need has been met by two books-Ostwald's "Physiko-Chemische Messungen" and Traube's "Physikalisch-Chemische Methode." Notwithstanding the excellence of these books they have not proven themselves practical guides in the laboratory, owing to too great detail and too many references to the literature. With the wish to prepare a manual which may be placed in the hands of the student of physical chemistry, the author has written this book. He would state at the outset that he in no way considers this as an effort to rival either of the above books. which for long must remain the standard works of reference on physico-chemical measurements. The effort has been made to select only such exercises as are typical, and where several different methods exist for the measurement of the same quantity, only in rare instances has more than one been given. In a word, the book has been made as condensed as possible in order not to discourage the student with too many methods.

It has been thought advisable to include several exercises which are usually studied in physics, but these may be omitted if the student has already had sufficient practice with them. For the convenience of the student there are appended tables of various physical constants which may be of service in the laboratory.

For the kind assistance given by friends the author would acknowledge his thanks, and especially to Prof. W. O. Atwater, who has so generously placed at his disposal the material for the section dealing with the determination of heats of combustion. To Messrs. P. Blakiston's Son & Co. I would express my thanks for permission to use several illustrations from Traube's "Physico-Chemical Methods."

If this book finds a place for itself in the hands of the student beginning the study of physical chemistry and proves to be of real service to him, it will have accomplished all that the author might wish.

FREDERICK H. GETMAN.

BALTIMORE, MD., April, 1904.

TABLE OF CONTENTS.

INTRODUCTORY MEASUREMENTS.

CHAPTER I.

STANDARD CONTRACTOR	AUI
Care of the Balance. Weighing by Vibrations. Sensitive- ness of a Balance. Inequality of the Arms of the Balance. Reduction of Weighing to Vacuo. Calibration of a Set of	
Weights.	
CHAPTER II.	
Volume and Density. Apparatus for Measuring Volumes. Calibration of Measuring-flasks. Calibration of Burettes. Calibration of a Eudiometer. Density (Specific Gravity). Density of Solids. Density of Liquids. Density of Gases (Vapor Density). Method of Victor Meyer (Air Displacement). Carrying out a Determination.	
CHAPTER III.	
Viscosity and Surface-Tension. Flow of Fluid through a Long Tube. Measurement of the Coefficient of Viscosity. Surface-tension. Measurement of Surface-tension.	31
CHAPTER IV.	
Solubility. Determination of Solubility.	4.5

THERMAL MEASUREMENTS.

8:3/9.2					
CH		TOTAL	20.70	17	77
	-			re ·	

	PAGE
THERMOMETRY. The Mercury Thermometer. Comparison of a Thermometer with a Standard Thermometer. Calibration by Means of a Series of Fixed Temperatures. Correction for Unheated Stem. The Fixed Points of a Thermometer. Expansion. Determination of the Coefficient of Cubical Expansion of Glass and Liquids. Molecular Volumes of Liquids at their Boiling-points.	
CHAPTER VI.	
Melting-point. Boiling-point. Depression of the Freezing-points of Solvents by Dissolved Substances. Apparatus and Method. Dissociation by the Freezing-point Method. Elevation of the Boiling-points of Solvents by Dissolved Substances. Apparatus and Method.	61
CHAPTER VII.	
Calorimetry. Quantity of Heat. Specific Heat. Determination of the Specific Heat of Solids. Heating Vessel. The Calorimeter. Method of Operation. Loss of Heat by Radiation. Loss of Heat to Calorimeter. Loss of Heat to Stirrer. Loss of Heat to the Thermometer. Determination of the Specific Heat of Liquids. Heat of Fusion. Heat of Vaporization. Thermochemistry. Heat of Neutralization. Heat of Solution. Heat of Hydration. Heat of Dilution. Heat of Combustion. Heat of Formation.	77
OPTICAL MEASUREMENTS.	
CHAPTER VIII.	
The Spectroscope	123

PAGE photometry. Apparatus. Method of Operation. Refractive Indices. Pulfrich Refractometer. Abbé Refractometer. Refraction Constants. The Polarimeter. Laurent Polarimeter. Lippich Polarimeter. Lamp for Homogeneous Light. Observing Tubes. Specific Rotation. Molecular Rotation. Rotation Dispersion. ELECTRICAL MEASUREMENTS. CHAPTER IX. Sources of Current. CHAPTER X. Specific and Molecular Conductivity. Resistance Boxes. Wheatstone's Bridge. Calibration of the Bridge-wire. Conductivity Cells. Inductorium and Telephone. Resistance Capacity of the Cell. Carrying out a Measurement. Pure Water. Equivalent Conductivity. Degree of Dissociation. The Dissociation Constant. The Basicity of Acids. Solubility by the Conductivity Method. CHAPTER XL Clark Standard Cell. Preparation of Materials. Temperature Coefficient. Weston Standard Cell. Helmholtz One-volt Cell. Lippmann Electrometer. The Measurement of Electromotive Force. Potential Differences. Normal Electrodes. Calibration of a Voltmeter. CHAPTER XII. The Silver Voltameter. Copper Voltameter. The Titration Voltameter. CHAPTER XIII.

Apparatus. Carrying out a Determination.

DYNAMICAL MEASUREMENTS.

CHAPTER XIV.

	PAGE
CHEMICAL KINETICS. Reaction of the First Order. Inversion of Cane-sugar.	212
Catalysis of Methyl Acetate. Reaction of the Second Order,	
Saponification of Ethyl Acetate.	
Tables	219
Reduction to Vacuum of Weighings made with Brass Weights	
in Air Density (Solids, Liquids, Gases). Density of Water.	
Volume of Water from 0° to 31°. Surface-tension of Liquids	
in Contact with Air. Viscosity of Liquids. Reduction of Gas	
Volumes to 0° and 760 mm. Reduction of Barometer Read-	
ings to 0°. Reduction of Mercury-in-Glass Thermometer	
Reading to the Normal Hydrogen Scale. Capillary Depression	
of Mercury. Vapor-pressure of Water. Vapor-pressure of	
Mercury. Table for Conversion of Thermometer Readings.	
Specific Heats, Heats of Fusion, and Melting-points of the	
Elements. Coefficients of Expansion, Specific Heats, Melting-	
points, and Boiling-points of Liquids. Boiling Temperatures	
of Water at Different Barometric Pressures. Correction for	
Temperature of Mercury in Thermometer Stem. Wave-lengths	
of Lines of Solar Spectrum in Air at 18°. Table for Wheatstone's	
Bridge. Table of International Atomic Weights. Table of	
Logarithms.	
Logarithms.	